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Chapters 17 (17.1 - 17.3): Instrumental Methods in Electrochemistry We use electrochemistry to do analytical chemistry in a variety of ways. We've already talked about pH electrodes and other kinds of ISE. Here are some more examples. Electrogravimetric Analysis The ____ part means we'll be using weighing as part of this

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Delahay, New Instrumental Methods in Electrochemistry (Interscience, New York, 1954), chap 8; W H Reinmuth, Anal Chem 32, 1514 (1960) 15 All potential values are with reference to the Ni/NiF₂ (1 mole percent) alkali fluoride reference electrode 16 Reduction of the pentavalent K₂NbF₇, the salt used in making up the solution, to the

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